**CORRECTION TO THE UNIT 6, PART 2 NOTES – AEROBIC CELLULAR RESPIRATION**

3) b. Cellular respiration is considered an oxidation-reduction reaction (aka a “Redox” reaction) because electrons are transferred from one molecule to another at many points throughout the process. The molecule that has lost the electron(s) is oxidized, and the molecule that has received the electron(s) is reduced. Usually, an H+ ion is lost or gained with the electrons, so the molecule that is oxidized will lose an H+, and the molecule that is reduced will gain an H+.